



Original Research Article

## Free radical-scavenging activity and cytotoxicity from leaves of *Guettarda platypoda*

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The specie *Guettarda platypoda* DC. a shrub present in the flora of the state of Pernambuco (Itamaracá) is used in folk medicine as an antipyretic. The aim of this study was to conduct a bioactivity-guided fractionation of the leaves of this species, thus evaluating the biological activity (Cytotoxic and antioxidant) of substance and/or fractions of different polarities. The raw methanol extract of the leaves of *G. platypoda* was subjected to fractionation using solvents of increasing polarity; cell viability was measured by MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) reduction assay; antioxidant activity was determined based on the ability to sequester free radicals by the method of DPPH (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) and the molecular characterization was performed by spectrometric methods. From this work it was possible to isolate a porphyrin (pheophytin a). During bioguided tests, it was evident that the crude methanolic extract ( $IC_{50} = 7.02 \times 10^{-3}$  mg/mL), fractions of ethyl acetate ( $IC_{50} = 5.58 \times 10^{-3}$  mg/mL) and methanol ( $IC_{50} = 8.10^{-3}$  mg/mL) had higher antioxidant activity than vitamin C. In conclusion, the substances present in the extract in ethyl acetate (F3) of *G. platypoda* and in the F2.2 subfraction, demonstrated cytotoxic and antioxidant activity against all cell lines tested, but the pheophytin a (P5), first identified to the genus *Guettarda*, showed no significant activity against the tested cell lines in this work.

**Key words:** *Guettarda platypoda*, phytochemical, pheophytina, cytotoxic activity, antioxidant.

### INTRODUCTION

The *Guettarda* genera consist of 180 species widely distributed in tropical and neotropical regions. Throughout the Brazilian territory there are about 24 species. There are reports of the use of this genre in traditional medicine for the most diverse purposes, especially for the treatment of wounds and inflammations (Agra et al., 2007; Agra et al., 2008; Bertucci et al., 2008; Capasso et al., 1998).

*Guettarda platypoda* DC is a shrubby species widely distributed along the Brazilian coast, especially in the Northeast (Pereira and Barbosa, 2004; Zickel et al., 2007). It is commonly used as antipyretic, anti-inflammatory

(Aquino et al., 1988a; Aquino et al., 1988b; Bhattacharyya and Almeida, 1985; Ferrari et al., 1986) and antiviral remedy against rhinovirus type 1B and the vesicular stomatitis virus (Aquino et al., 1989).

An examination of the extract of the bark of the root of the plant verified the presence of saponines, alkaloids, steroids, tannins and triterpenes, as well as the lack of flavonoids. The quinovic acid, rotundic acid,  $\beta$ -sitosterol (Almeida, 2011), the 5 $\alpha$ -carboxistrictosidina alkaloid, iridoid morroniside the secoiridoid sweroside (Ferrari et al., 1986), triterpene saponins, loganic acid, loganin

(Aquino et al., 1988a; Aquino et al., 1988b) and pentacyclic triterpenes (Soares et al., 1998) were isolated.

In the methanol extract of the leaves of *G. platypoda* was identified the presence of flavonoids, tannins (proanthocyanidins condensed leucoanthocyanidins) saponoside, triterpenes and steroids and the absence of alkaloids, coumarins, iridoids, as well as chlorogenic acid (cinnamic acid derivative), rutin, quercetin and kempferol (flavonoids) (Corrêa, 2007).

Given this variety of compounds and the few reports on the species in question, the present study aimed to conduct a guided fractionation and evaluate the cytotoxic and antioxidant activities of the crude methanol extract and subfractions of the leaves of the *G. platypoda* DC.

## MATERIALS AND METHODS

### Collection, Identification and Preparation of Extract

The plant material (leaves) was collected in the municipality of Itamaracá, Pernambuco, about 50 km from Recife, bounded on the north by the municipality of Goiana, on the south by Igarassu to the east by the Atlantic Ocean and to the west by Itapissuma (Zickel et al., 2007). A specimen of the species was identified and deposited in the Herbarium of the Agronomic Institute of Pernambuco (IPA) under number 86565.

After collection, the leaves were dried using an oven at 48 °C for 48 hours to eliminate moisture and stabilize the enzyme content. The material was powdered in electric mill and weighed (736.8 g). The extraction of the leaves of the fixed components was carried out by maceration for seven days using methanol PA (Merck) (20 g/L). The extracts were filtered and the residue re-extracted. The solutions obtained in this process were concentrated at reduced pressure in a rotary evaporator at  $50 \pm 5$  °C until the solvent had been completely removed. The result was 173.3 g of the crude methanol extract (CME).

### Fractionation bioassay-guided

The CME leaves of *G. platypoda* (8 g) was subjected to fractionation in a column using Traditional Liquid Chromatography (TLC) with silica gel 60 (0.063-0.2 mm, 70-230 mesh, Macherey-Nagel) as the stationary phase and with solvents PA (Merck) of increasing polarity. This procedure resulted in four fractions: *n*-hexane (F1), *n*-hexane/ethyl acetate (AcOEt) 1:1 (F2), AcOEt (F3) and methanol MeOH (F4) that were subsequently concentrated under reduced pressure in a rotary evaporator. The fractions were evaluated for their antioxidant and cytotoxic activities, determining their therapeutic potential by directing the fractionation (Houghton, 1996).

Then the F2 and F3 fractions were subjected to fractionation in TCL with silica gel to obtain subfractions

F2.1, F2.2 and P5, respectively. The P5 subfraction led to the isolation of a substance (Pheophytin a) with a quantity and pure quality sufficient for analysis in Nuclear Magnetic Resonance (NMR) 1D  $^1\text{H}$  (500 MHz) and  $^{13}\text{C}$  (125 MHz).

### Cytotoxicity assay

Cell viability was measured by MTT reduction assay, which is based on the conversion of the salt 3-(4,5-dimethyl-2-thiazole)-2,5-diphenyl-2H-tetrazolium bromide (MTT) to a formazan product by the action of the enzyme succinyl dehydrogenase present in mitochondria of viable cells. The analysis used the CME leaves of *G. platypoda*, its fractions and pure compound isolated against the strains HEP-2 (human epidermoid Cancer Cells; HeLa derivative), HT-29 (colon carcinoma), MCF-7 (human breast adenocarcinoma cell line) and HL-60 (leukemia), each a concentration of 50g/mL. The cancer cell lines DMEM (Dulbecco's Modification of Eagle's Medium) were plated in 96-well microplates and incubated during 24 h at 37 °C. After this period, 25  $\mu\text{L}$  of CME leaves of *G. platypoda* and its fractions (50  $\mu\text{g}/\text{mL}$ ) were added to each well and incubated for 72 h. Next, MTT dye (25  $\mu\text{L}$ , 0.5 mg/mL) was added to the wells and the assay was incubated for 3 h more. After this period the medium was removed and dimethylsulfoxide (DMSO, 100  $\mu\text{L}$ ) was added to the wells for solubilization of the generated formazan salts. The optical density of the wells was measured at 540 nm and compared to that in the negative control (cells incubated only with medium). Two independent experiments were performed in duplicate. The samples were classified according to the inhibition percent of cell viability in the following categories: inactive (1-20%), weakly active (20-50%), moderately active (50-70%) or very active (70-100%) (Capasso et al., 1998).

### Antioxidant activity test

The antioxidant activity was determined based on the ability of scavenge free radicals, similar to that described by Hegazi et al. (2002), using the 2,2-Diphenyl-1-(2,4,6-trinitrophenyl)hydrazino (DPPH). Methanolic solutions of the CME and the fractions obtained at concentrations of 0.001; 0.005; 0.01; 0.05; 0.1 and 1 mg/mL were prepared. An aliquot (1.0 mL) of each solution was mixed with another methanolic DPPH solution (1.0 mL) at a concentration of 60  $\mu\text{mol}\cdot\text{L}^{-1}$ . After 30 minutes the absorbance measurements in a UV-VIS spectrophotometer at 520 nm were carried out. The percentage of inhibition was obtained by comparing the absorption of the solution containing the sample, relative to a control DPPH solution without sample. The test was performed in triplicate, using as Trolox as the positive standards at concentrations of 0.001; 0.002; 0.004; 0.005; 0.1 and 1 mg/mL and vitamin C at 0.0025; 0.005; 0.0125; 0.025; 0.0375 and 0.050 mg/mL. The calculation of the antioxidant activity was made using

GraphPad Prism 4 for Windows program and the calculation of IC<sub>50</sub> (concentration which inhibits 50% of free radicals) used the sigmoidal dose-response equation.

### Molecular characterization

The molecular characterization was performed by spectrometric methods such as infrared spectrum (IR) and NMR (Bruker Daltonics) 1D <sup>1</sup>H (500 MHz) and <sup>13</sup>C (125 MHz) and homonuclear 2D <sup>1</sup>H-<sup>1</sup>H COSY (COrrrelation Spectroscopy) and heteronuclear <sup>13</sup>C-<sup>1</sup>H HSQC (Heteronuclear Single Quantum Coherence) and HMBC (Heteronuclear Multiple Bond Correlation)

## RESULTS AND DISCUSSIONS

### Fractionation bioassay-guided

The chromatographic fractionation of 8 g of crude methanolic extract of the leaf of *G. platypoda* (EBMFG) was obtained through a column filled with silica gel powder with 70-230 mesh which was eluted with different solvents in order of increasing polarity. From this procedure four fractions were obtained: F1-*n*-hexane (0.13 g); F2-*n*-hexane/AcOEt [1:1] (0.55 g); F3-AcOEt (0.16 g); F4-methanol (4.5 g).

Then F3 fraction was subjected to TLC with silica gel, eluting with mixtures of hexane and AcOEt in increasing order of polarity. The P5 subfraction resulting from this process was possible to isolate a green amorphous solid, which after NMR (Tables 1 and 2) and infrared analysis and comparisons with similar procedures in the literature (Aquino et al., 1998a; Aquino et al., 1988b), it was possible to identify as pheophytin a (Figure 1), a porphyrin previously not reported for this genus.

### Molecular characterization

The Infrared spectrum of pheophytin a (IR (KBr): 1698; 1738; 2926; 2847; 3434 cm<sup>-1</sup>) showed bands attributed to conjugated carbonyl group ( $\nu_{\max}$  1698 cm<sup>-1</sup>), ester function ( $\nu_{\max}$  1738 cm<sup>-1</sup>), aliphatic and aromatic C-H ( $\nu_{\max}$  2926, 2847 cm<sup>-1</sup>) and N-H amines ( $\nu_{\max}$  3434 cm<sup>-1</sup>), along with characteristics of double bonds of alkenes and aromatic rings (Aquino et al., 1998a).

Comparative analysis of the {<sup>1</sup>H}- and DEPT <sup>13</sup>C NMR spectra allowed us to identify signals (Tables 1 and 2) corresponding to nineteen non-hydrogenated carbons [all sp<sup>2</sup>]: including three carbonyl groups at  $\delta_c$  189.87 (C-13<sup>1</sup>), 169.83 (C-13<sup>3</sup>) and 173.17 (C-17<sup>3</sup>). Eleven methine carbons (including five sp<sup>2</sup>) were identified. Fourteen methylene carbons [including one sp<sup>2</sup> at  $\delta_c$  122.96 (CH<sub>2</sub>-3<sup>2</sup>) and one sp<sup>3</sup> oxygenated at  $\delta_c$  61.67 (CH<sub>2</sub>-3<sup>1</sup>)] were identified. Eleven methyl carbons [including one methoxyl group at  $\delta_c$  53.04 (MeO-13<sup>3</sup> = 3H<sub>3</sub>-13<sup>4</sup>). All of these were

used to build the partial molecular formula (C)<sub>16</sub>(C=O)<sub>3</sub>(CH)<sub>11</sub>(CH<sub>2</sub>)<sub>14</sub>(CH<sub>3</sub>) (Aquino et al., 1989).

The <sup>1</sup>H NMR spectra (1D and 2D <sup>1</sup>H-<sup>1</sup>H-COSY) of pheophytin a showed singlet signals at  $\delta_H$  9.56, 9.35 and 8.56, attributed to characteristic hydrogen H-10, H-5 and H-20, respectively, of the presence of porphyrin moiety by corresponding to the carbon methinics bound to pyrrole rings. Signals were also observed corresponding to five methyl groups at  $\delta_H$  3.69 (s, Me-12<sup>1</sup>), 3.40 (s, Me-2<sup>1</sup>); 3.21 (s, Me-7<sup>1</sup>), 1.82 (d, *J*=7.0 Hz, Me-18<sup>1</sup>) and 1.69 (t, *J*=7.5 Hz, Me-8<sup>2</sup>) appearing as substituents (four methyl and one ethyl-8 groups) on the pyrrole rings of the porphyrin unit (Table 1) (Aquino et al., 1998a).

Further data was found to confirm the presence of a porphyrin moiety: a triplet signal at  $\delta_H$  3.66 (q, *J*=7.5 Hz, 2H-8<sup>1</sup>), methoxyl group at  $\delta_H$  3.90 (br s, 3H-13<sup>4</sup>) and a vinyl substituent linked to C-3 by signals  $\delta_H$  7.97, [dd, 17.8 and 11.5 Hz, H-3<sup>1</sup> and 6.29 (d, *J*=17.8 Hz, H3<sup>2a</sup>) and 6.14 (d, *J*=11.5 Hz, H3<sup>2b</sup>) correlated in the HSQC with carbon signals at  $\delta_c$  129.28 (CH-3<sup>1</sup>) and 122.96 (CH<sub>2</sub>-3<sup>2</sup>), along with additional signals to methines CH-13<sup>2</sup>. [ $\delta_H$  6.28(s)/  $\delta_c$  64.91], CH-17 [ $\delta_H$  4.25(m)/  $\delta_c$  51.83] and CH-18 [ $\delta_H$  4.45(m)/ $\delta_c$  50.33] and methylenes CH<sub>2</sub>-17<sup>1</sup> [ $\delta_H$  2.64(m), 2.37(m)/ $\delta_c$  30.03] and CH<sub>2</sub>-17<sup>2</sup> [ $\delta_H$  2.50(m), 2.23(m)/ $\delta_c$  31.45], consistent with porphyrin ring (Zickel et al., 2007).

According Schwikkard et al. (1998), the pheophytin b in C-7<sup>1</sup> has an aldehyde group, differing from the compound found in this study due to the presence of a methyl signal at  $\delta_H$  3.21, reinforcing the suggestion that it is pheophytin a (Table 1).

The long range heteronuclear coupling of the carbonyl C-17<sup>3</sup> ( $\delta_c$  173.17) with 2H-17<sup>1</sup> ( $\delta_H$  2.64 and 2.37, <sup>3</sup>J<sub>CH</sub>), 2H-17<sup>2</sup> ( $\delta_H$  2.50 and 2.23, <sup>2</sup>J<sub>CH</sub>) and 2H-1<sup>1</sup> ( $\delta_H$  4.57 and 4.44, <sup>3</sup>J<sub>CH</sub>) was used to confirm the esterification of the carboxyl group of the porphyrin moiety. All chemical shift assignments of the hydrogen and carbon atoms remaining of this moiety were also attributed on the basis of 1D and 2D NMR spectra (Table 1).

The attribution of the chemical shifts of the hydrogen and carbon atoms of the phytyl group (Table 2) was also based in the 1D and 2D NMR spectral data involving comparison with values described in literature (Capasso et al., 1998; Zickel et al., 2007).

Thus, <sup>1</sup>H and <sup>13</sup>C NMR spectral data, involving 1D and 2D [<sup>1</sup>H-<sup>1</sup>H-COSY and <sup>1</sup>H-<sup>13</sup>C-COSY, HSQC, and HMBC correlations] and comparison with procedures reported in the literature were used to characterize and to allow the complete and unambiguous chemical shift assignments (Tables 1 and 2) of pheophytina.

### Cytotoxic activity

Fouche et al, (2008) asserted that extracts tested at a maximum single dose of 100 µg/mL exhibit cytotoxic activity since they are able to inhibit the proliferation of

**Table 1.**  $^1\text{H}$  (500 MHz) and  $^{13}\text{C}$  (125 MHz) spectral data of porphyrin unit, in  $\text{CDCl}_3$ , inclusive results of HSQC and HMBC. Chemical shifts in  $\delta$  (ppm).

| C               | HSQC                |  | HMBC                     |   |
|-----------------|---------------------|--|--------------------------|---|
|                 | $\delta_{\text{C}}$ | $\delta_{\text{H}}$                            | $^2\text{J}_{\text{CH}}$ | $^3\text{J}_{\text{CH}}$                    |
| 1               | 142.27              | -  |                          | 3H-21                                       |
| 2               | 132.06              | -  | 3H-21                    | H-2; H-31                                   |
| 3               | 136.45              | -  | H-31                     | 2H-32                                       |
| 4               | 136.74              | -  |                          | H-31  |
| 6               | 155.85              | -  |                          | 3H-71                                       |
| 7               | 136.38              | -  | 3H-7 <sup>1</sup>        | H-5; 2H-8 <sup>1</sup>                      |
| 8               | 145.49              | -  | 2H-8 <sup>1</sup>        | H-10; 3H-7 <sup>1</sup> ; 3H-8 <sup>2</sup> |
| 9               | 151.19              | -  |                          | 2H-8'                                       |
| 11              | 138.14              | -  | H-10                     | 3H-12 <sup>1</sup>                          |
| 12              | 129.30              | -  |                          | H-10  |
| 13              | 129.19              | -  |                          | H-13 <sup>2</sup> ; 3H-12 <sup>1</sup>      |
| 14              | 149.85              | -  |                          | H-13 <sup>2</sup>                           |
| 15              | 105.46              | -  | H-13 <sup>2</sup>        |   |
| 16              | 161.48              | -  |                          |   |
| 19              | 172.45              | -  | H-18                     | 3H-18 <sup>1</sup>                          |
| 13 <sup>1</sup> | 189.87              | -  | H-13 <sup>2</sup>        |   |
| 13 <sup>3</sup> | 169.83              | -  | H-13 <sup>2</sup>        | MeO-13 <sup>4</sup>                         |
| 17 <sup>3</sup> | 173.17              | -  | 2H-17 <sup>2</sup>       | 2H-1 <sup>1</sup> ; 2H-17 <sup>1</sup>      |
| CH              |                     |  |                          |   |
| 5               | 97.75               | 9.35 (s)                                       |                          |   |
| 10              | 104.63              | 9.56 (s)                                       |                          |   |
| 17              | 51.83               | 4.25 (m)                                       |                          | 3H-18 <sup>1</sup>                          |
| 18              | 50.33               | 4.45   | 3H-18 <sup>1</sup>       |   |
| 20              | 93.33               | 8.56 (s)                                       |                          |   |
| 3 <sup>1</sup>  | 129.28              | 7.97 dd, J 17.8, 11.5                          |                          |   |
| 13 <sup>2</sup> | 64.91               | 6.28 (s)                                       |                          |   |
| CH2             |                     |  |                          |   |
| 3 <sup>2</sup>  | 122.96              | 6.29 (d, J = 17.8 Hz)<br>6.14 (d, J = 11.5 Hz) |                          |   |
| 8 <sup>1</sup>  | 19.63               | 3.66 (q)                                       | 3H-82                    |   |
| 17 <sup>1</sup> | 30.03               | 2.64 (m), 2.37 (m)                             |                          |   |
| 17 <sup>2</sup> | 31.45               | 2.50 (m), 2.23 (m)                             |                          |   |
| CH3             |                     |  |                          |   |
| 2 <sup>1</sup>  | 12.28               | 3.40 (s)                                       |                          |   |
| 7 <sup>1</sup>  | 11.40               | 3.21 (s)                                       |                          |   |
| 8 <sup>2</sup>  | 17.59               | 1.69 (t, J = 7.5 Hz)                           | 2H-81                    |   |
| 12 <sup>1</sup> | 12.28               | 3.69 (s)                                       |                          |   |
| 13 <sup>4</sup> | 53.04               | 3.90 (s)                                       |                          |   |
| 18 <sup>1</sup> | 23.29               | 1.82 (d, J = 7.3 Hz)                           | H-18                     |   |

two or more cell lines by at least 75%. Based on this information, the F2.2 and F3 fractions exhibit cytotoxic activity against the HEp-2 and HT-29 (Table 3) strains. However, the pheophytin, showed no significant results.

The presence of terpenoids can justify the cytotoxic action of these two fractions when an MTT assay was studied. These secondary metabolites are well known for their antitumor activity. Between 1980 and 2000 they the main representative of the paclitaxel category, considered at that time to be the most promising antitumor agents

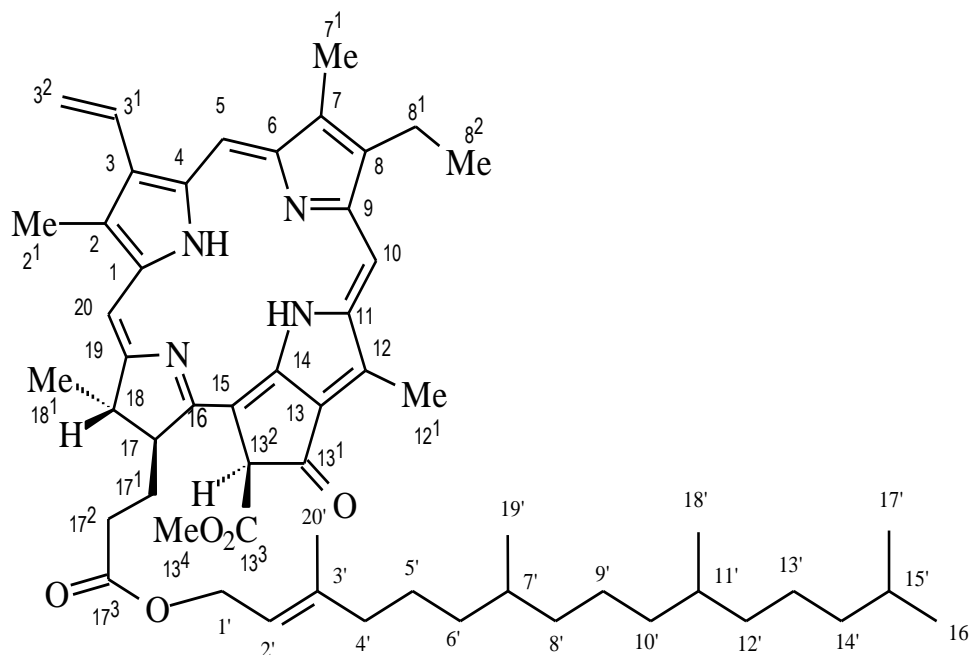
(Simões et al., 2010).

### Antioxidant activity

The CME leaves of *G. platypoda*, their primary pheophytin a and the fractions were analyzed. The results showed that both the CMB as F3 and F4 fractions exhibited antioxidant activity with 99% inhibition of free radicals in a concentration of 0.1 mg/mL. The IC<sub>50</sub> was less than the standard used (vitamin C) has proven to have high

**Table 2.**  $^1\text{H}$  (500 MHz) and  $^{13}\text{C}$  (125 MHz) spectral data of **phytyl group**, in  $\text{CDCl}_3$ , inclusive results of HSQC and HMBC. Chemical shifts in  $\delta$  (ppm).

|                 | HSQC                |                          | HMBC                     |                          |
|-----------------|---------------------|--------------------------|--------------------------|--------------------------|
|                 | $\delta_{\text{C}}$ | $\delta_{\text{H}}$      | $^2\text{J}_{\text{CH}}$ | $^3\text{J}_{\text{CH}}$ |
| C               |                     |                          |                          |                          |
| 3'              | 143.02              | -                        | 3H-20'/ 2H-4'/           | 2H-1'/ 2H-5'             |
| CH              |                     |                          |                          |                          |
| 2'              | 117.94              | 5.15 (t, $J = 6.9$ Hz)   | 2H-1''                   | 3H-20'                   |
| 7'              | 32.96               | 1.31 (m)                 | 3H-19'                   |                          |
| 11'             | 32.81               | 1.31 (m)                 | 3H-18'                   |                          |
| 15'             | 28.11               | 1.51 (hep, $J = 6.6$ Hz) | 3H-16'/3H-17'            |                          |
| CH <sub>2</sub> |                     |                          |                          |                          |
| 1'              | 61.67               | 4.52 (m), 4.44 (m)       |                          |                          |
| 4'              | 39.99               | 1.89 (t, $J = 7.7$ Hz)   |                          | 3H-20'                   |
| 5'              | 25.19               | 1.61 (m)                 |                          |                          |
| 6'              | 36.83               | 1.02 (m)                 |                          | 3H-19'                   |
| 8'              | 37.59               | 1.21 (m)                 |                          | 3H-19'                   |
| 9'              | 24.61               | 1.12 (m)                 |                          |                          |
| 10'             | 37.53               | 1.23 (m)                 |                          | 3H-18'                   |
| 12'             | 37.46               | 1.23 (m)                 |                          | 3H-18'                   |
| 13'             | 24.98               | 1.27 (m)                 |                          |                          |
| 14'             | 39.55               | 1.12 (m)                 |                          |                          |
| CH <sup>3</sup> |                     |                          |                          |                          |
| 16'             | 22.91               | 0.86 (d, $J = 6.5$ Hz)   | H-15'                    |                          |
| 17'             | 22.81               | 0.86 (d, $J = 6.5$ Hz)   | H-15'                    |                          |
| 18'             | 19.91c              | 0.81 (d, $J = 6.6$ Hz)   |                          |                          |
| 19'             | 19.63c              | 0.80 (d, $J = 6.6$ Hz)   |                          |                          |
| 20'             | 17.59               | 1.59 (sl)                |                          | H-2'                     |

**Figure 1:** The structure of pheophytin

antioxidant activity (Mensor et al., 2001) (Table 4). This effect is probably due to the presence of flavonoids <sup>14</sup> in fraction F3 and flavonoids, tannins and reducing sugars in

F4 and CME, secondary metabolites known to have antioxidant activity (Simões et al., 2010).

In their studies Endo et al. (1984, 1985) reported that the

**Table 3.** Evaluation of the cytotoxic activity of the crude methanol extract of the leaves of *G. platypoda* and its fractions.

| Samples                     | HEp-2 | HT-29 | MCF-7 | HL-60 |
|-----------------------------|-------|-------|-------|-------|
| Methanolic fraction (F4)    | 25.2  | 31.3  | -     | -     |
| Ethyl acetate fraction (F3) | 92.8  | 77.8  | -     | -     |
| F2.1                        | 80.5  | 38.4  | -     | -     |
| F2.2                        | 99.7  | 86.7  | -     | -     |
| N-hexane fraction (F1)      | 40.3  | 35.5  | -     | -     |
| CME                         | 40.7  | 53.8  | -     | -     |
| Doxorubicine* (15 µg/mL)    | 0.7   | 0.4   | -     | 0.02  |
| Pheophytin a (P5)           | -     | -     | 25.49 | 26.79 |

Percentage of growth inhibition from crude methanol extract (CME) of the leaves of *G. platypoda*; (F1) *n*-hexane fraction, (F2.1 and F2.2) *n*-hexane/ ethyl acetate (1:1); (F3) ethyl acetate fraction, (F4) methanolic fraction; and \* substances standart. (-) no active.

**Table 4.** Evaluation of antioxidant activity of the crude methanol extract of the leaves of *G. platypoda* and its fractions.

| Amostras     | IC <sub>50</sub> (mg/mL)                     |
|--------------|--|
| CME          | 7.02.10 <sup>-3</sup> ± 2.8.10 <sup>-5</sup> |
| F1           | -  |
| F2           | -  |
| F3           | 5.58.10 <sup>-3</sup> ± 7.0.10 <sup>-4</sup> |
| F4           | 8.10 <sup>-3</sup> ± 1.1.10 <sup>-5</sup>    |
| Pheophytin a | 7.79.10 <sup>-3</sup> ± 2.2.10 <sup>-3</sup> |
| Trolox *     | 2.6.10 <sup>-3</sup> ± 2.3.10 <sup>-4</sup>  |
| Vitamin C *  | 8.14.10 <sup>-3</sup> ± 5.9.10 <sup>-4</sup> |

(CME) crude methanol extract of the leaves of *G. platypoda*; (F1) *n*-hexane fraction, (F2) *n*-hexane/ ethyl acetate (1:1); (F3) the ethyl acetate fraction, (F4) methanolic fraction; and \* substances standart.

chlorophyll and the pheophytins can move from prooxidants when subjected to light, catalyzing oxidation reactions and antioxidant, under the shelter of light. Thus, as expected, the pheophytin showed antioxidant activity, since the experiment was always protected from light.

## Conclusions

In conclusion, the substances presents in fraction F3 (AcOEt) and F2.2 subfraction of *G. platypoda* demonstrated cytotoxic and antioxidant activity against all cell lines tested, but the pheophytin a, first identified to the genus *Guettarda*, showed no significant activity against the tested cell lines in this work.

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